

OLI Software Release Notes V12.x

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Introduction

Each version will be listed in reverse chronological order. Within each version update to the products will be listed in this order:

- Databank updates
- Databank fixes
- Engine/solver
- OLI Studio/Analyzer
- OLI Flowsheet: ESP
- OLI Chemistry Wizard
- OLI Developer Edition
- OLI Alliance products
 - Aspen Hysys
 - o Aspen Plus
 - o gProms
 - o IDEAS
 - o Petro-Sim
 - o Proll
 - o Unisim Design
- OLI Security/License Manager

Version 12.0.1.11

General Information

Version 12 of the OLI Software was officially released on May 29, 2024

Product	Release Date
OLI Studio/Analyzer (x64)	May 29, 2024
OLI Flowsheet: ESP / ESP FS ¹ (x64)	May 29, 2024
Chemistry Wizard	May 29, 2024
OLI Developer Edition (x64)	May 29, 2024
OLI Engine 12.0.1.11 for Aspen Hysys (x64 bit)	May 29, 2024
OLI Engine 12.0.1.11 for Aspen Plus (x64 bit)	May 29, 2024
OLI Engine 12.0.1.11 for Petro-Sim (x64 bit)	May 29, 2024
OLI Engine 12.0.1.11 for ProII (x64)	May 29, 2024
OLI Engine 12.0.1.11 for Unisim Design	May 29, 2024
OLI License Manager ² (x64 bit)	May 29, 2024

¹ The product ESP FS is for Asia releases. It is functionally identical to OLI Flowsheet: ESP.

² This product does not follow the same versioning scheme as the other products. The current version is 5.5.0.10

Databank Updates

Mixed-solvent electrolyte (MSE) databank thermophysical properties

Complexes between Fe and organic species

Fe (II, III) – acetate Fe (II, III) – diethylenetriamine Fe (II, III) – ethylenediamine Fe (II, III) – glycine Fe (II, III) – pyrozole Fe (II, III) – catechol Fe (II, III) – gallic acid Fe (II, III) – humic acid Fe (II, III) – phenol

Butyric acid revision, including Ca and K butyrate and butyric + propionic acids

Butyric acid - H₂O (revised) Butyric acid - propionic acid - H₂O (revised) Calcium butanoate - H₂O (revised) Potassium butanoate - H₂O (revised)

$NH_3 + K_2CO_3 + MDEA$ and related systems

 $\begin{array}{l} NH_3 - K_2CO_3 - H_2O \ (revised) \\ NH_3 - K_2CO_3 - (NH_4)_2CO_3 - H_2O \ (revised) \\ MDEA - K_2CO_3 - H_2O \\ MDEA - NH_3 - K_2CO_3 - H_2O \\ MDEA - CO_2 - K_2CO_3 - H_2O \\ MDEA - K_2CO_3 - CO_2 - NH_3 - H_2O \\ MDEA - NH_3 - H_2O \\ MDEA - O_2 - NH_3 - H_2O \\ MDEA - NH_4HCO_3 - H_2O \\ MDEA - NH_3 - NH_4HCO_3 - H_2O \\ \end{array}$

Piperazine + CO_2 + H_2S + H_2O + MDEA

 $\begin{array}{l} Piperazine-H_2O\\ Piperazine-CO_2-H_2O\\ Piperazine-MDEA-H_2O\\ Piperazine-CO_2-MDEA-H_2O\\ Piperazine-H_2S-H_2O\\ Piperazine-H_2S-MDEA-H_2O\\ \end{array}$

Revision of properties of selenium species: Se⁰ and SeO₄²⁻ Se – NaCl – H₂O

Crystalline calcium silicates moved from CRMSE databank to GEMSE

Afwillite, $Ca_3Si_2O_4(OH)_6$ Foshagite, $Ca_4Si_3O_9(OH)_2.0.5H_2O$ Gyrolite, $Ca_2Si_3O_7(OH)_2.1.5H_2O$ Hatrurite, Ca_3SiO_5 Hillebrandite, $Ca_2SiO_3(OH)_2.0.17H_2O$ Okenite, $Ca_3Si_2O_4(OH)_2.H_2O$ Rankinite, $Ca_3Si_2O_7$ Tobermorite11A, $Ca_5Si_6H_{11}O_{22.5}$ Tobermorite14A, $Ca_5Si_6H_{21}O_{27.5}$ Tobermorite9A, $Ca_5Si_6H_6O_{20}$ Xonotlite, $Ca_6Si_6O_{17}(OH)_2$ Clinochrysotile, Mg₃Si₂O₅(OH)₄

Cobalt chemistry: CoSO4 (NH4)2SO4, Ni_{0.78}Co_{0.22}(NH4)2(SO4)2, Co-NH3 complexes

 $\begin{array}{l} (NH_4)_2SO_4 - CoSO_4 - H_2O\\ CoSO_4(NH_4)_2SO_4 - H_2O\\ NiSO_4(NH_4)_2SO_4 - CoSO_4(NH_4)_2SO_4 - H_2O\\ System \ where \ Ni_{0.78}Co_{0.22}(SO_4)_2(NH_4)_2\cdot 6H_2O \ precipitates \end{array}$

$MnSO_4 + (NH_4)_2SO_4 + H_2O$

 $MnSO_4 - (NH_4)_2SO_4 - H_2O$

Nd citrate entropy and Cp: correction

Al phosphates

 $AlPO_4 - H_3PO_4 - H_2O$ $Al_2O_3 - P_2O_5 - H_2O$

NH₄ phosphates

 $\begin{array}{l} NH_4H_2HPO_4-NH_3-H_3PO_4-H_2O\\ (NH_4)_2HPO_4-NH_3-H_3PO_4-H_2O\\ (NH_4)_2HPO_4\cdot 2H_2O-NH_3-H_3PO_4-H_2O\\ (NH_4)_3PO_4\cdot 3H_2O-NH_3-H_3PO_4-H_2O\\ \end{array}$

Mn phosphates

 $MnO - P_2O_5 - H_2O$ $Mn_3(PO_4)_2 - H_3PO_4 - H_2O$

Fe phosphates

 $\begin{array}{l} FePO_4 - H_2O\\ FePO_4 - H_3PO_4 - H_2O \end{array}$

Zn phosphates

 $Zn_3(PO_4)_2-H_3PO_4-H2O$

$Na_2HPO_4 + NaNO_3 + H_2O$

 $Na_2HPO_4 - NaNO_3 - H_2O$

H₃BO₃ + sulfates (Na, K, Li, Mg, Ca)

 $\begin{array}{l} H_{3}BO_{3}-Na_{2}SO_{4}-H_{2}O\\ H_{3}BO_{3}-K_{2}SO_{4}-H_{2}O\\ H_{3}BO_{3}-Li_{2}SO_{4}-H_{2}O\\ H_{3}BO_{3}-MgSO_{4}-H_{2}O\\ H_{3}BO_{3}-CaSO_{4}-H_{2}O\\ HBO_{2}\ (aq,\ vap)\ removed\ -\ only\ HBO_{2}(s)\ remains \end{array}$

Rare earth element hydroxycarbonates, carbonates and REE fluorocarbonates

 $\begin{array}{l} YOHCO_3 - (CO_2) - H_2O\\ LaOHCO_3 - (CO_2) - H_2O\\ CeOHCO_3 - (CO_2) - H_2O\\ PrOHCO_3 - (CO_2) - H_2O\\ NdOHCO_3 - (CO_2) - H_2O\\ SmOHCO_3 - (CO_2) - H_2O\\ EuOHCO_3 - (CO_2) - H_2O\\ \end{array}$

$Guomeo_3 - (CO_2) - m_2O$
$TbOHCO_3 - (CO_2) - H_2O$
$DyOHCO_3 - (CO_2) - H_2O$
$HoOHCO_3 - (CO_2) - H_2O$
$ErOHCO_3 - (CO_2) - H_2O$
$TmOHCO_3 - (CO_2) - H_2O$
YbOHCO ₃ – (CO_2) – H ₂ O
$L_{11}OHCO_3 - (CO_2) - H_2O$
$Y_2(CO_3)_3 - H_2O_1$
$La_2(CO_3)_3 - H_2O_1$
$Ce_2(CO_3)_3 - H_2O_2$
$\frac{Pr_2(CO_2)_2}{Pr_2(CO_2)_2} = H_2O$
$Nd_2(CO_2)_2 - H_2O_2$
$Sm_2(CO_3)_2 - H_2O_1$
$Fu_2(CO_2)_2 = H_2O_1$
$Gd_2(CO_2)_2 - H_2O_2$
$Tb_2(CO_2)_2 - H_2O_2$
$D_{V_2}(CO_2)_2 = H_2O$
$H_{02}(CO_3)_3 - H_2O_1$
$\operatorname{Ho}_2(\operatorname{CO}_3)_3 = \operatorname{H}_2O$ $\operatorname{Fre}_2(\operatorname{CO}_2)_2 = \operatorname{H}_2O$
$\operatorname{Er}_{2}(\operatorname{CO}_{3})_{3} = \operatorname{Er}_{2}\operatorname{CO}_{3}$ $\operatorname{Er}_{2}(\operatorname{CO}_{3})_{3} = \operatorname{Er}_{2}\operatorname{CO}_{3}$
$\operatorname{Hil}_2(\operatorname{CO}_3)_3 = \operatorname{Hil}_2(\operatorname{CO}_3)_3 $
$10_2(CO_3)_3 - 11_2O_1$
$Lu_2(CO_3)_3 - H_2O_3$
$VFCO_2 - H_2O_2$
$I_{2}ECO_{2} = H_{2}O_{2}$
$C_{a}ECO_{a} = H_{2}O$
$\mathbf{PrECO}_{2} = \mathbf{H}_{2}\mathbf{O}$
$HFCO_3 - H_2O_3$
$\operatorname{NuFCO}_{3} = \operatorname{H_2O}_{2O}$
$SIIIFCO_3 - \Pi_2 O$
$Eurco_3 - H_2O$
$Ourco_3 - n_2O$
$10FCO_3 - H_2O_2$
$DyFCO_3 - H_2O$
$\Pi 0 \Gamma C O_3 - \Pi_2 O$
$ErFCO_3 - H_2O$
$ErFCO_3 - H_2O$ TmFCO ₃ - H ₂ O
$ \begin{array}{l} ErFCO_3 - H_2O\\ TmFCO_3 - H_2O\\ YbFCO_3 - H_2O\\ \end{array} $
$\begin{array}{l} ErFCO_3-H_2O\\ TmFCO_3-H_2O\\ YbFCO_3-H_2O\\ LuFCO_3-H_2O \end{array}$
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O Rare earth element oxyfluorides and oxychlorides
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O Rare earth element oxyfluorides and oxychlorides YOCl – H ₂ O
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O Rare earth element oxyfluorides and oxychlorides YOCl – H ₂ O LaOCl – H ₂ O
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O Rare earth element oxyfluorides and oxychlorides YOCl – H ₂ O LaOCl – H ₂ O CeOCl – H ₂ O
$ \begin{array}{l} ErFCO_3 - H_2O\\ TmFCO_3 - H_2O\\ YbFCO_3 - H_2O\\ LuFCO_3 - H_2O\\ \end{array} \\ \hline {\it Rare \ earth \ element \ oxyfluorides \ and \ oxychlorides}\\ YOCl - H_2O\\ LaOCl - H_2O\\ CeOCl - H_2O\\ PrOCl - H_2O\\ \end{array} \\ \hline {\it NlOCl - H_2O}\\ \hline {\it NlOCl - H_2O}\\ \end{array} $
$ \begin{array}{l} ErFCO_3 - H_2O\\ TmFCO_3 - H_2O\\ YbFCO_3 - H_2O\\ LuFCO_3 - H_2O\\ \end{array} \\ \hline {\it Rare \ earth \ element \ oxyfluorides \ and \ oxychlorides}\\ YOCl - H_2O\\ LaOCl - H_2O\\ CeOCl - H_2O\\ PrOCl - H_2O\\ NdOCl - H_2O\\ \end{array} $
$ErFCO_3 - H_2O$ $TmFCO_3 - H_2O$ $YbFCO_3 - H_2O$ $LuFCO_3 - H_2O$ $Rare earth element oxyfluorides and oxychlorides$ $YOC1 - H_2O$ $LaOC1 - H_2O$ $CeOC1 - H_2O$ $PrOC1 - H_2O$ $NdOC1 - H_2O$ $SmOC1 - H_2O$
$\begin{aligned} & \text{ErFCO}_3 - \text{H}_2\text{O} \\ & \text{TmFCO}_3 - \text{H}_2\text{O} \\ & \text{YbFCO}_3 - \text{H}_2\text{O} \\ & \text{LuFCO}_3 - \text{H}_2\text{O} \\ & \text{\textbf{Bare earth element oxyfluorides and oxychlorides}} \\ & \text{YOCI} - \text{H}_2\text{O} \\ & \text{LaOCI} - \text{H}_2\text{O} \\ & \text{LaOCI} - \text{H}_2\text{O} \\ & \text{CeOCI} - \text{H}_2\text{O} \\ & \text{PrOCI} - \text{H}_2\text{O} \\ & \text{NdOCI} - \text{H}_2\text{O} \\ & \text{SmOCI} - \text{H}_2\text{O} \\ & \text{EuOCI} - \text{H}_2\text{O} \\ & \text{EuOCI} - \text{H}_2\text{O} \end{aligned}$
$ErFCO_3 - H_2O$ $TmFCO_3 - H_2O$ $YbFCO_3 - H_2O$ $LuFCO_3 - H_2O$ $Rare earth element oxyfluorides and oxychlorides$ $YOCI - H_2O$ $LaOCI - H_2O$ $CeOCI - H_2O$ $PrOCI - H_2O$ $NdOCI - H_2O$ $SmOCI - H_2O$ $EuOCI - H_2O$ $EuOCI - H_2O$
$ErFCO_3 - H_2O$ $TmFCO_3 - H_2O$ $YbFCO_3 - H_2O$ $LuFCO_3 - H_2O$ $Rare earth element oxyfluorides and oxychlorides$ $YOCI - H_2O$ $LaOCI - H_2O$ $CeOCI - H_2O$ $PrOCI - H_2O$ $NdOCI - H_2O$ $SmOCI - H_2O$ $EuOCI - H_2O$
$ \begin{array}{l} {\rm ErFCO_3-H_2O} \\ {\rm TmFCO_3-H_2O} \\ {\rm YbFCO_3-H_2O} \\ {\rm LuFCO_3-H_2O} \\ {\rm LuFCO_3-H_2O} \\ \\ {\rm XOCl-H_2O} \\ {\rm LaOCl-H_2O} \\ {\rm CeOCl-H_2O} \\ {\rm PrOCl-H_2O} \\ {\rm NdOCl-H_2O} \\ {\rm NdOCl-H_2O} \\ {\rm SmOCl-H_2O} \\ \\ {\rm SmOCl-H_2O} \\ \\ {\rm EuOCl-H_2O} \\ \\ {\rm EuOCl-H_2O} \\ \\ {\rm DyOCl-H_2O} \\ \\ {\rm H_2O} \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\ \\$
$\begin{aligned} & \text{ErFCO}_3 - \text{H}_2\text{O} \\ & \text{TmFCO}_3 - \text{H}_2\text{O} \\ & \text{YbFCO}_3 - \text{H}_2\text{O} \\ & \text{LuFCO}_3 - \text{H}_2\text{O} \\ & \text{LuFCO}_3 - \text{H}_2\text{O} \\ & \text{Rare earth element oxyfluorides and oxychlorides} \\ & \text{YOCl} - \text{H}_2\text{O} \\ & \text{LaOCl} - \text{H}_2\text{O} \\ & \text{CeOCl} - \text{H}_2\text{O} \\ & \text{PrOCl} - \text{H}_2\text{O} \\ & \text{NdOCl} - \text{H}_2\text{O} \\ & \text{SmOCl} - \text{H}_2\text{O} \\ & \text{EuOCl} - \text{H}_2\text{O} \\ & \text{EuOCl} - \text{H}_2\text{O} \\ & \text{GdOCl} - \text{H}_2\text{O} \\ & \text{CbOCl} - \text{H}_2\text{O} \\ & \text{TbOCl} - \text{TbOCL} \\ & \text{TbOCL} \\ & \text{TbOCL} - \text{TbOCL} \\ & \text{TbOCL} \\ & T$
$ErFCO_3 - H_2O$ $TmFCO_3 - H_2O$ $YbFCO_3 - H_2O$ $LuFCO_3 - H_2O$ $Rare earth element oxyfluorides and oxychlorides$ $YOC1 - H_2O$ $LaOC1 - H_2O$ $CeOC1 - H_2O$ $PrOC1 - H_2O$ $NdOC1 - H_2O$ $SmOC1 - H_2O$ $EuOC1 - H_2O$ $EuOC1 - H_2O$ $EuOC1 - H_2O$ $FOOC1 - H_2O$
ErFCO ₃ – H ₂ O TmFCO ₃ – H ₂ O YbFCO ₃ – H ₂ O LuFCO ₃ – H ₂ O $Rare \ earth \ element \ oxyfluorides \ and \ oxychlorides$ YOCl – H ₂ O LaOCl – H ₂ O CeOCl – H ₂ O PrOCl – H ₂ O NdOCl – H ₂ O SmOCl – H ₂ O EuOCl – H ₂ O GdOCl – H ₂ O TbOCl – H ₂ O DyOCl – H ₂ O TbOCl – H ₂ O CeOCl – H ₂ O TbOCl – H ₂ O TbOCl – H ₂ O TbOCl – H ₂ O
$ \begin{array}{l} \mbox{ErFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{TmFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{YbFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{LuFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{LuFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{LuFCO}_3 - \mbox{H}_2\mbox{O}\\ \mbox{VoCl} - \mbox{H}_2\mbox{O}\\ \mbox{LaOCl} - \mbox{H}_2\mbox{O}\\ \mbox{CeOCl} - \mbox{H}_2\mbox{O}\\ \mbox{CeOCl} - \mbox{H}_2\mbox{O}\\ \mbox{SmOCl} - \mbox{H}_2\mbox{O}\\ \mbox{SmOCl} - \mbox{H}_2\mbox{O}\\ \mbox{EuOCl} - \mbox{H}_2\mbox{EuOCl} - \mbox{H}_2\mbox{EuOCl} - \mbox{H}_2\mbox{EuOCl} - \mbox{H}_2\mbox{EuOCl} - \mbox{EuOCl} - \mbox{H}_2\mbox{EuOCl} - \mbox{EuOCl} - \m$

$YOF - H_2O$
$LaOF - H_2O$
$CeOF - H_2O$
$PrOF - H_2O$
$NdOF - H_2O$
$SmOF - H_2O$
$EuOF - H_2O$
$GdOF - H_2O$
$TbOF - H_2O$
DyOF – H ₂ O
HoOF – H ₂ O
$ErOF - H_2O$
$TmOF - H_2O$
$YbOF - H_2O$
LuOF – H ₂ O

Crystalline rare earth hydroxides moved from MSEPUB to GEMSE and CRMSE

 $\begin{array}{l} Y(OH)_{3}cr\\ La(OH)_{3}cr\\ Ce(OH)_{3}cr\\ Pr(OH)_{3}cr\\ Nd(OH)_{3}cr\\ Sm(OH)_{3}cr\\ Sm(OH)_{3}cr\\ Gd(OH)_{3}cr\\ Gd(OH)_{3}cr\\ Tb(OH)_{3}cr\\ Ho(OH)_{3}cr\\ Er(OH)_{3}cr\\ Tm(OH)_{3}cr\\ Yb(OH)_{3}cr\\ Lu(OH)_{3}cr\\ \end{array}$

Lithium chemistry $(Li + Na + K + Cl + SO_4 + H_2O)$ (revision)

$$\begin{split} &Na_2SO_4 - Li_2SO_4 - H_2O\\ &NaCl - LiCl - H_2O\\ &Li_2SO_4 - Na_2SO_4 - K_2SO_4 - H_2O\\ &LiNO_3 - KNO_3 - H_2O\\ &Li_2SO_4 - K_2SO_4 - H_2O\\ &LiCl - KCl - H_2O\\ &Li_2SO_4 - H_2O \ density\\ &Li_2SO_4 - LiCl - H_2O\\ &Li_2SO_4 - K_2SO_4 - H_2O\\ &Li_2SO_4 - MgSO_4 - H_2O\\ &Li_2SO_4 (LiCl) - CaSO_4(CaCl_2) - H_2O\\ &Li_2SO_4 - MnSO_4 - H_2O\\ &Li_2SO_4 - (NH_4)_2SO_4 - H_2O\\ &Li_2SO_4 - NiSO_4 - H_2O\\ &Li_2SO_4 - NiSO_4 - H_2O\\ \end{split}$$

Co hydrolyzed species – conventional and non-conventional species

 $\begin{array}{l} Co(OH)_2-H_2O\\ Co(OH)_2-NaOH(KOH)-H_2O\\ Co(OH)_2-HCl-H_2O \end{array}$

 $EuO + H_2O$

 $EuO-NaOH-H_2O\\$

*LiBF*₄ + organic carbonates

LiBF₄ – Dimethyl carbonate (Lithium tetrafluoroborate hemidimethyl carbonate, Lithium tetrafluoroborate dimethyl carbonate)

Co and Ni solvent extraction chemistry

CYANEX272-H₂O CYANEX301-H₂O $DEHPA - H_2O$ $PC88A - H_2O$ VERSATIC10-H₂O Co and Ni extraction using CYANEX272 extractant and isooctane as solvent Co and Ni extraction using CYANEX272 extractant and high boiling kerosene (KEROSENE) as solvent Co and Ni extraction using CYANEX272 extractant and low boiling kerosene (KEROSENUN) as solvent in the presence of Tributyl phosphate (TRIBUTPHOS) Co and Ni extraction using CYANEX272 extractant and Toluene as solvent Co extraction using CYANEX272 extractant and n-Heptane as solvent Co extraction using PC88A extractant and n-Heptane as solvent Co and Ni extraction using D2EHPA extractant and low boiling kerosene (KEROSENUN) as solvent in the presence of Tributyl phosphate (TRIBUTPHOS) Li and Co extraction using D2EHPA extractant and high boiling kerosene (KEROSENE) as solvent Li, Co, and Ni extraction using PC88A extractant and high boiling kerosene (KEROSENE) as solvent Li, Co, and Ni extraction using PC88A extractant and low boiling kerosene (KEROSENUN) as solvent Li, Co, and Ni extraction using PC88A extractant and low boiling kerosene (KEROSENUN) as solvent in the presence of Tributyl phosphate (TRIBUTPHOS)

MnCO₃ and Mn(ClO₄)₂ systems MnCO₃ – NaCl – NaClO₄ – Na₂CO₃ – H₂O

 $MnCO_3 - NaCl - NaClO_4 - HCl - H_2O$

*MgCl*₂ – *adding anhydrous species* MgCl₂ – H₂O

KOH revision + related systems

$$\begin{split} & \text{KOH} - \text{H}_2\text{O} \text{ (revised)} \\ & \text{KOH} - \text{Ag}_2\text{O} - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{Be}(\text{OH})_2 \ / \ \text{BeO} - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{CH}_4 - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{GaOOH} - \text{H}_2\text{O} \text{ including potassium gallate (tested)} \\ & \text{KOH} - \text{GaOOH} - \text{H}_2\text{O} \text{ including potassium borates (tested)} \\ & \text{KOH} - \text{H}_3\text{BO}_3 - \text{H}_2\text{O} \text{ including potassium borates (tested)} \\ & \text{KOH} - \text{H}_3\text{BO}_3 - \text{H}_2\text{O} \text{ including potassium borates (tested)} \\ & \text{KOH} - \text{Np}(\text{IV}) - \text{K}_2\text{CO}_3 (\text{KHCO}_3) - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{PbCO}_3 - \text{CO}_2 (\text{KHCO}_3) - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{Pu}(\text{IV}) - \text{K}_2\text{CO}_3 (\text{KHCO}_3) - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{tannic acid} - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{U}(\text{IV}) - \text{K}_2\text{CO}_3(\text{KHCO}_3) - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} - \text{ZnO} - \text{H}_2\text{O} \text{ (tested)} \\ & \text{KOH} (\text{K}_2\text{O}) - \text{B}_2\text{O}_3 - \text{H}_2\text{O} \text{ (revised)} \\ & \text{KOH} (\text{K}_2\text{O}) - \text{B}_2\text{O}_3 - \text{H}_2\text{O} \text{ (revised)} \\ & \text{KOH} - \text{Zn}(\text{OH})_2 \ / \ \text{ZnO} - \text{H}_2\text{O} \text{ (revised)} \end{split}$$

Density of $NaHCO_3 + H_2O$ and $NaHCO_3 + Na_2CO_3 + H_2O$

 $\begin{array}{l} NaHCO_3-H_2O\\ NaHCO_3-Na_2CO_3-H_2O \end{array}$

 $KBr + H_2O$

 $KBr-H_{2}O \\$

$Ca(OH)_2 + NaOH and LiOH + H_2O$

 $\begin{array}{l} Ca(OH)_2 - NaOH - H_2O \\ Ca(OH)_2 - LiOH - H_2O \end{array}$

$Cl_2 - HCl - chloride \ salts$

Mixed-solvent electrolyte (MSE) databank transport properties

(revised, electrical conductivity)
(revised, electrical conductivity)
(thermal conductivity)
(revised, viscosity)
(revised, electrical conductivity)
(revised, electrical conductivity)
(thermal conductivity)
(revised, viscosity)
(electrical conductivity)

Mixed-solvent electrolyte (MSE) databank Scaling inhibition kinetics

SrSO ₄	(revised)
BaSO ₄	(revised)
CaSO4 [·] 2H ₂ O	(revised)
CaCO ₃ (Calcite)	(revised)
SrSO4-HEDP	(revised)
SrSO4-DTPMP	(revised)
BaSO4-HEDP	(revised)
BaSO4-NTMP	(revised)
BaSO4-DTPMP	(revised)
CaSO4 [·] 2H ₂ O-HEDP	(revised)
CaSO4 [·] 2H ₂ O-NTMP	(revised)
CaSO4 [·] 2H ₂ O-DTPMP	(revised)
CaSO4 [·] 2H ₂ O-EDTMP	
CaSO ₄ ·2H ₂ O-PBTC	(revised)
CaCO ₃ (Calcite)-HEDP CaCO ₃ (Calcite)-NTMP CaCO ₃ (Calcite)-DTPMF CaCO ₃ (Calcite)-PMLA	(revised) (revised) (revised) (revised)

HEDP: 1-hydroxyethane 1,1-diphosphonic acid NTMP: Nitrilotris(methylenephosphoric acid) DTPMP: Diethylenetriamine penta(methylene phosphonic acid) EDTMP: Ethylenediamine tetra(methylene phosphonic acid) PBTC: 2-phosphono-butane-1,2,4-tricarboxylic acid PMLA: Poly maleic acid

MSE Corrosion: Electrochemical kinetics databank

MSE Alloy 2507

Corrosive environments: NaCl $Seawater - O_2$ $MgCl_2 - NaCl$ $NaCl - O_2$ $H_2S - NaCl$ $CO_2 - NaCl$ $CO_2 - H_2S - NaCl$ $NaCl - acetic acid - H_2S - CO_2$ NaOH NaOH – NaCl Formic acid Acetic acid Formic acid - acetic acid H_2SO_4 HC1 HNO₃ H₃PO₄ $HCl - H_2SO_4$ $HCl - HNO_3$ HF $H_2SO_4 - HNO_3$ $HCl - H_3PO_4$ $HNO_3 - H_3PO_4$ $H_2SO_4 - HF$ $\mathrm{HF}-\mathrm{HNO}_3$ $HF - H_3PO_4$ HCl - HF $HBr - H_3PO_4$ H₂SO₄ - acetic acid H₂SO₄-formic acid NaCl - acetic acid NaCl-formic acid $CuCl_2$ FeCl₃ $FeCl_3-FeCl_2-NaCl-HCl\\$ $CuCl_2-CuCl-NaCl-HCl \\$ $H_2SO_4 - Fe_2(SO_4)_3$ H₂SO₄-Fe₂(SO₄)₃-NaCl

Repassivation potential parameters:

Cl⁻ H₂S SO₄²⁻ NO3⁻ OH⁻ VO4³⁻ MoO4²⁻ NO2⁻

MSE Alloy 2205

Corrosive environments: NaCl $Seawater-O_2 \\$ $MgCl_2 - NaCl$ $NaCl - O_2$ $H_2S - NaCl$ $CO_2 - NaCl$ $CO_2-H_2S-NaCl \\$ $NaCl-acetic \; acid-H_2S-CO_2$ NaOH NaOH – NaCl Formic acid Acetic acid Formic acid - acetic acid H_2SO_4 HCl HNO₃ H₃PO₄ $HCl - H_2SO_4$ $HCl - HNO_3$ HF $H_2SO_4-HNO_3\\$ $HCl - H_3PO_4$ $HNO_3 - H_3PO_4$ $H_2SO_4 - HF$ $\mathrm{HF}-\mathrm{HNO}_3$ $\mathrm{HF}-\mathrm{H_3PO_4}$ HCl – HF $HBr - H_3PO_4$ H₂SO₄ – acetic acid H₂SO₄-formic acid NaCl – acetic acid NaCl-formic acid CuCl₂ FeCl₃ $FeCl_3 - FeCl_2 - NaCl - HCl \\$ $CuCl_2-CuCl-NaCl-HCl \\$ $H_2SO_4 - Fe_2(SO_4)_3$ $H_2SO_4 - Fe_2(SO_4)_3 - NaCl$

Repassivation potential parameters:

Cl⁻ H₂S SO₄²⁻ NO₃⁻ OH⁻ VO₄³⁻ MoO4²⁻ NO2⁻

Aqueous (AQ) thermophysical property databank

Crystalline calcium silicates moved from the CORROSION databank to GEOCHEM:

Afwillite, $Ca_3Si_2O_4(OH)_6$ Foshagite, $Ca_4Si_3O_9(OH)_2.0.5H_2O$ Gyrolite, $Ca_2Si_3O_7(OH)_2.1.5H_2O$ Hatrurite, Ca_3SiO_5 Hillebrandite, $Ca_2SiO_3(OH)_2.0.17H_2O$ Okenite, $Ca_3Si_2O_4(OH)_2.H_2O$ Rankinite, $Ca_3Si_2O_7$ Tobermorite11A, $Ca_5Si_6H_{11}O_{22.5}$ Tobermorite14A, $Ca_5Si_6H_{21}O_{27.5}$ Tobermorite9A, $Ca_5Si_6H_2O_{20}$ Xonotlite, $Ca_6Si_6O_{17}(OH)_2$ Clinochrysotile, $Mg_3Si_2O_5(OH)_4$

AQ Corrosion: Electrochemical kinetics databank

AQ Alloy 304 – revision

Corrosive environments:

 $\begin{array}{c} O_2 \\ NaCl - O_2 \\ Seawater - O_2 \\ MgCl_2 - O_2 \\ Na_2SO_4 - O_2 \\ NaCl - HF - O_2 \\ HF - O_2 \\ H_2O_2 - H_2 - O_2 \\ H_2O_2 - H_2 - O_2 - NaCl - FeCl_3 \\ Cl_2 - O_2 \\ NaOH - O_2 \\ LiOH - O_2 \\ B(OH)_3 - O_2 \end{array}$

AQ Alloy 316 – revision

Corrosive environments: NaCl HCl HCl - HNO₃ HCl - H₃PO₄ HCl - H₂SO₄ FeCl₃ FeCl₂ CuCl₂ CuCl

AQ Alloy 2205 – revision

Corrosive environments:

 $\begin{array}{l} O_2 \\ HF \\ Seawater - O_2 \\ NaCl - O_2 \\ NaOH- O_2 \end{array}$

AQ Alloy 2507 – revision

Corrosive environments:

 $\begin{array}{l} O_2 \\ HF \\ NaCl-NaF-O_2 \\ Seawater-O_2 \\ NaCl-O_2 \\ NaOH-O_2 \\ H_3PO_4 - HF \end{array}$

OLI Databook – Bugs & Features & Resolved issues.

Bug ID	Problem	Resolution
DT-2	Electrical Conductivity Data missing for HCIO	Update data for V12
DT-428	Rare Earth hydroxides with the formula RE(OH)3 not recognized	Code updated.
DT-441	Eliminate the name "Celestite" from SrSO4 name	Removed.
DT-459	Some chemical formulas have square brackets (e.g., La[C2H3O2]3 instead of parenthesis. This prevents proper database searches by chemical formula	290 species were updated.
DT-528	Add piperazine to the MSE database (already in AQ)	Added.
DT-547	Searching for a species not in the database creates a pseudo-species	Updated the code to protect against invalid searches.
DT-598	Add MnCO3 to MSE (already in AQ)	Added.
DT-708	The synonym for TI(OH) is incorrect. It is marked as TI(NO3)	Updated
DT-722	Add Co-NH3 complexes to MSE	Added
DT-756	Add Zn3(PO4)2 to MSE	Added
DT-829	Data for Ca(OH)2-NaOH-H2O is under predicting	Updated
DT-830	Data for LiOH-Ca(OH)2-H2O is not regressed	Updated

Engine/Solver - Features

Bug ID	AREA	Problem	Resolution
		No new features were added for version 12.0.1.11	

Engine/Solver - Bugs

Bug ID	AREA	Problem	Resolution
DT-675	Solver	Incorrect phase behavior was found in a carbon capture case	There is an odd phase transition in the CO2 VLE. Special phase check handlers were added to detect this special condition.

DT-695	Solver	A Flowsheet case fails with multiple models with ion- exchange.	The second model did not properly update all variables which caused the error.
DT-750	Solver	Reactor did not mass balance with user entered reaction kinetics	There was an issue with the chemistry model parsing the equation. This was corrected.
DT-791	Solver	A mixer case in OLI Studio worked in previous versions but failed in 11.5	The case requires a better estimate of initial temperature in adiabatic cases. Handlers were added to provide better estimates.

OLI Studio/Analyzer - Features

Bug ID	AREA	Problem	Resolution
DT-386	Scaling	A user cannot easily identify a scaling inhibitor in the	The Letter "I" has been added to the grid to show scaling
	Inhibition	inflow list	inhibition chemicals.
DT-426	Chemistry	The database lists were not sorted by the display name but rather by the file name	List updated to be sorted by display name
DT-838	Kinetics Editor	No reaction kinetics editor exists in survey calculations	This feature was added.

OLI Studio/Analyzer - Bugs

Bug ID	AREA	Problem	Resolution
DT-534	Report	Selected species to report not honored, all species displayed.	Logic updated.
DT-556	ScaleChem Report	Exporting a facility calculation to a CSV file has a maximum of 14 nodes.	This restriction has been removed.
DT-564	Chemistry Model	Redox subsystem missing tag in UI	Subsystem not properly tagged in database
DT-565	Chemistry Model	Redox subsystem for Pb (AQ framework) missing an oxidation state.	Subsystem not properly tagged in AQ PUBLIC database.
DT-574	UI	Copying data from an inflow grid to another file crashed the program.	A deadlock in the UI was found and resolved.
DT-575	Chemistry Model	The interaction matrix did not report transport properties (e.g., viscosity)	The calls to the generator were updated to include transport properties
DT-740	Report	The program crashed when clicking on the report tab.	A private database was used in this case that did not have the MATC records completely filled out. The report code was enhanced to trap this condition and display blanks.
DT-764	Chemistry Model	A user entered reaction kinetic parameter had a stray character. This caused the program crash when running.	Stray characters which are normally permitted in FORTRAN equations, e.g., "*", are now trapped and a warning provided.
DT-803	UI	Using gauge pressure (BARG or PSIG) was entered as "0" but returned a very small non-zero value after calculation, sometimes negative	Updated logic properly displays gauge pressure.
DT-813	UI	A case saved in a previous version with custom calculations enabled shows a blank screen when opened with the option not enabled	Custom calculations are now enabled by default.
DT-923	UI	Water Analysis in MSE using expanded species results in invalid species (e.g., HION)	Expanded species updated for MSE

OLI Flowsheet: ESP / ESP-FS – Features/Improvements

Bug ID	AREA	Problem	Resolution
DT-54	UI	There is a limit of 50 controllers	Limit increased to 100. Older files that are opened in V12 may display a momentary error which will resolve itself.
DT-84	UI	The convergence monitor drop-down menu is to narrow.	Field was widened.
DT-291	Chemistry	OLI Studio has the ability to modify the critical parameters of a pseudocomponent from an assay. This ability does not exit in OLI Flowsheet: ESP	Added support to modify generated pseudocomponent data from an assay.
DT-417	Chemistry/UI	A user cannot identify a private database name from the chemistry model view. This feature exists in OLI Studio	The UI was modified to show the internal name of the database in a similar fashion as OLI Studio.
DT-420	UI	Sensitivity plots do not include block properties such as heat duty.	Block results are now captured for each sensitivity run.

DT-434	Chemistry	Add classical inert species as found in OLI Chemistry Wizard	Species added.
DT-482	UI	The error panel column is not adjustable.	The columns are now adjustable.
DT-491	UI	The term "Dissolved Phases" is incorrect in a separator	Changed to "Entrained"
DT-611	UI	To phase separate the results of a chemical reactor a user must also use a separator unit operation.	The reactor and separator blocks have now been combined.
DT-646	UI/Units Manager	°F and °C units are missing in heat exchanges when U*A is specified. Only absolute K is permitted	Units added.
DT-670	Sensitivity	The sensitivity study does not work for parameters in a multi-stage block (a/k/a column)	Parameters were added for the sensitivity study.

OLI Flowsheet: ESP / ESP-FS - Bug

Bug ID	AREA	Problem	Resolution
DT-127	Optimizer	An optimized final value was negative, but the report shows a positive value	The report captured a value just before the final iteration of the optimizer. The logic was updated to capture the final value.
DT-377	Chemistry/Reactor	A reaction rate does not have a reverse reaction (KR=0) yet the report shows a reverse reaction.	An very small value was entered to indicate a zero reaction (1E-300). This is now explicitly made zero in the report.
DT-515	Sensitivity	Sensitivity calculation reports errors about species missing in the chemistry model.	The chemistry model removed species which were monitored or adjusted in the sensitivity block. Logic was enhanced to warn about such situations.
DT-523	Chemistry/Reactor	Reaction Kinetics does not work with ion-exchange reactions	The OLI Reaction Kinetics was not designed to work with SOL species. For this version, the reactor is allowing to run without kinetics till a permanent solution is found.
DT-578	Units Manager	ScaleChem units are appearing in OLI Flowsheet	ScaleChem units have been removed.
DT-584	Sensitivity	The range (when decremented) was not being honored	The range is now honored.
DT-586	UI	The call out for a flow split reports flows as mole % when the actual splitter units were in mass %	Units are now being honored.
DT-591	UI/Report	Pre-scaling tendencies are enabled but do not show up in all streams that have solids eligible for pre-scaling calculations.	The restart option did not always save post-calculation variables such as pre-scaling tendencies. The logic was updated to save this information and other post- calculations values (e.g., TDS, Alkalinity, etc.)
DT-657	UI	Opened report tabs are not saved when reopening a saved file	Logic enhanced to remember which tabs are open when being saved.
DT-676	Units Manager	For MSE the units manager should use Liquid-1 instead of Aqueous	Table was updated.
DT-758	Chemistry Model	Error when entering a species from the drop-down menu	The species was previously entered but deleted. The UI did not understand that the species had been removed. The logic was updated.
DT-889	Solver/UI	Dew Point calculation does not work when calculating the dew point pressure	The block was originally set to bubble point pressure but then changed to dew point pressure. The UI did not update the call to the solver to indicate that it is now a dew point.
DT-893	Solver	An adiabatic mixer has a different phase split (VLE) than an isothermal calculation at the same converged temperature.	The adiabatic mixer used a poor initialized temperature. Other changes to the solver fixed the initialized variable.
DT-926	Solver	The case file, previously saved in an earlier version, fails to open with two private database	The memory stack was exceeded due to new features having been added. The stack was reduced in size.

Alliance Products - Features

Bug ID	Program	Problem	Resolution
		No new features were added for version 12.0.1.11	

Alliance Products - Bugs

Bug ID	Program	Problem	Resolution
DT-725	Unisim	Case crashes with S(0) redox enabled.	OLI Engine updated to prevent crash.
	Design		

OLI Chemistry Wizard (all products)

Bug ID	Program	Problem	Resolution
DT-771	All	Oil assay fails to generate with then name HSP2	Cross-referencing name logic was updated.

OLI Developer Edition - General

Bug ID	Program	Problem	Resolution
		No new features were added for version 12.0.1.11	

OLI Framework - General

Bug ID	Program	Problem	Resolution
		No new features were added for version 12.0.1.11	

OLI License Manager / Security

Bug ID	Program	Problem	Resolution
DT-582	License	Trailing space at the end of the server name in the OLI	Trailing spaces trimmed
	Manager	Software is not recognized as a valid name	
DT-887	License	Previous log files are autodeleted when rolling over to a	Previous log files are not deleted.
	Manager	new file	